

Si₃N₄/Mg composites with an interpenetrating network

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Introduction

Metal matrix composites (MMCs) have some kinds of reinforcement such as particle, whisker and fibre [1–3] as well as three dimensional network structure (3DNMMCs) [4]. In recent years, the MMCs reinforced by 3DNMMCs have been received much attention [5]. Especially, 3DNMMCs reinforced by ceramic phase with an interconnected interpenetrating network structure have become of great interest because of their attractive microstructure and properties [6, 7]. For example, the Si₃N₄ ceramic with network structure interpenetrating ductile metal magnesium would take an amazing effects in microstructure and properties due to the ceramic phase providing low density, high elastic modulus and high strength while the metallic phase offering high toughness as well as high thermal and electrical conductivity [8, 9]. Such composites can be produced by infiltration of a molten metal into porous ceramic skeleton under or without pressure. Porous ceramic skeleton can be fabricated by a variety of methods such as replication technique, gas bubble formation and space holder methods [10, 11]. The most common method of making an open cell ceramic foam skeleton is replication

technique, which has been known for almost 40 years (which was first developed in the early 1960s) [12]. Polymer foam of desired high open pore is immersed into a low viscosity ceramic slurry, thereby burn out and finally sintered [10, 13]. The infiltration technique is the most attractive processing route to fabricate 3DNMMCs which can be classified into three categories based on the source of driving force: pressure assisted, vacuum driven and pressureless or capillarity driven [14]. These methods cause a great contact area between the components [14]. During this process the interfacial reactions between metal and ceramic can take place, which leads to the appearance of a spinel phase on the ceramic/metal interface. The interface characteristic decides about the properties of the composites.

In this paper, the authors attempt to attribute the microstructure of porous ceramic skeleton and increase investigations of the ceramic/metal interface where products of a reaction between the liquid metal and the ceramics can appear. Preliminary results of studies of Si₃N₄/Mg composites obtained via infiltration of a porous Si₃N₄ ceramic skeleton are reported.

Experimental procedure

Fabrication of porous ceramic skeleton

A commercial β -Si₃N₄ powder (Si₃N₄ \geq 97%, diameter \leq 100 μ m, supplied by Shanghai Silicon Materials Plant, China) was used as starting material to fabricate the porous ceramic skeleton. The sintering additives as 5 wt%ZrO₂ and 2 wt%Al₂O₃ (Al₂O₃ \geq 99.66%, diameter \leq 90 μ m, supplied by Shandong Aluminum Industry Company, China) and 5 wt%Al (Al \geq 99.26%, D₅₀ = 4 \pm 1.5 μ m, supplied by

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Shandong Aluminum Industry Company, China) were mixed with the starting materials and ball-milled for 4 h using Al_2O_3 balls. A commercial silica sol (SiO_2 , 30.0–31.0%; pH, 8.5–10.0, supplied by Shanghai Kangning Silica Sol Company, China) was used as a binder to be added in the resulting powder mixture. A kind of surfactant carboxymethyl cellulose was added to serve as a wetting agent. The commercial polyurethane sponges (PUS, supplied by Shanghai Nanjia Sponges Production Company, China) with reticulated structure, interconnected struts and open porosity of approximately 10 pores per inch (ppi) were chosen in this study. The cleared PUS was cut into standard form of ϕ 30×100 mm and immersed into the homogeneous slurry for about 10 min. The samples were placed into the drying oven under 160°C for at least 20 h. The sintering process was carried through at 1400°C for 2 h in oxidizing atmosphere and followed by cooling in the furnace to room temperature. The representative images of the macroscopic structure of porous reticulated ceramic were presented in Fig. 1. All these samples were selected as a potential ceramic phase to the experiments of infiltration by magnesium metal.

Fabrication of $\text{Si}_3\text{N}_4/\text{Mg}$ composites

The chemical compositions for the alloy (which was shown in Table 1) for the infiltration were prepared in a resistance furnace under a protective atmosphere of N_2 (99.996%) to avoid oxidation. Pure Al (99.99%), Mg (99.96%), and Mg-3%Bi alloy were used as raw materials to prepare the alloys. The preforms were heated in N_2 at a rate of $15^\circ\text{C}/\text{min}$ up to the test temperature. After achieving the infiltration temperature, the liquid metal magnesium was infiltrated into the preform skeleton based on capillarity driven. Two specimens (B and C) at different process parameters as infiltration temperature (T), infiltration time (t_I) and heat hold time (t_H) were fabricated. Those different processes lead to different interface characteristics and properties such as compressive strength (σ_C), prolongation (L) and tensile strength (σ_T) (as shown in Table 2,

Table 1 The chemical compositions for magnesium alloy, wt%

Mg	Bi	Zn	Al	Si	Ti	Ni	Cu
90.328	0.108	1.132	8.33	0.04	0.043	0.014	0.005

Table 2 The infiltration processes and properties of alloy and 3DNMMCs

Specimen	T ($^\circ\text{C}$)	t_I (min)	t_H (min)	σ_C (MPa)	L (%)	σ_T (MPa)
A	–	–	–	240	5	230
B	1050	150	180	250	2	241
C	920	60	60	255	2	49

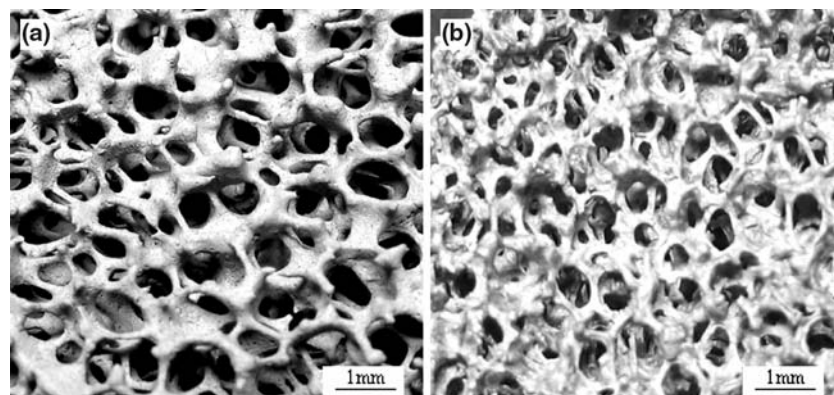
specimen A as the alloy). The micro-structural characteristic of $\text{Si}_3\text{N}_4/\text{Mg}$ composites with a three-dimensional interpenetrating network was performed on a scanning electron microscopy (SEM, HITACHI, Model No. S-2500) which presented in Fig. 2.

Results and discussion

Microstructure of the porous ceramic skeleton

The relative density of porous ceramic was measured by using the Archimedes' method. The pore size was determined by the bubble method [15]. In this method the pore size is defined by the pressure required for the first air bubble to be pressed out by the porous sample. The width of strut (t), the length of strut (l) (which was shown in Fig. 3) and the number of pore were also determined by the microstructure of porous ceramic skeleton such as Fig. 4. In order to obtain relatively accurate results, the method of statistical treatment is necessary to be used. The cross-section and surface micro-structural characterization of the samples were performed on a SEM (HITACHI, Model No. S-2500). The obtained results are presented in Table 3. The relative density was validated by the following equation [16]:

Fig. 1 Porous reticulated ceramic structure. (a) Before sintering (b) after sintering at 1400°C



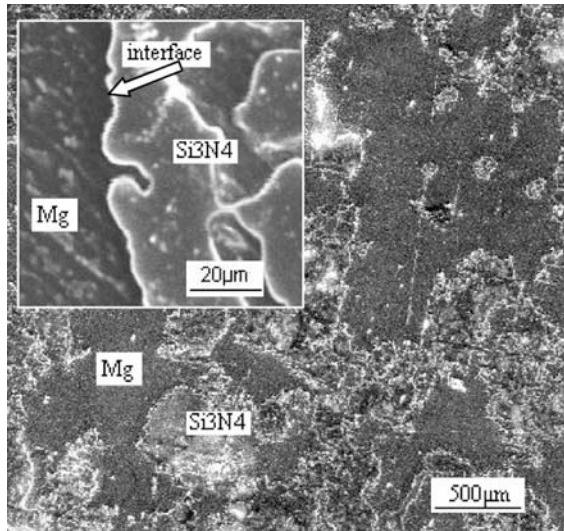


Fig. 2 The microstructure of Si₃N₄/Mg composites with an interpenetrating network. The bright phase is Si₃N₄ ceramic skeleton, the dark phase is magnesium

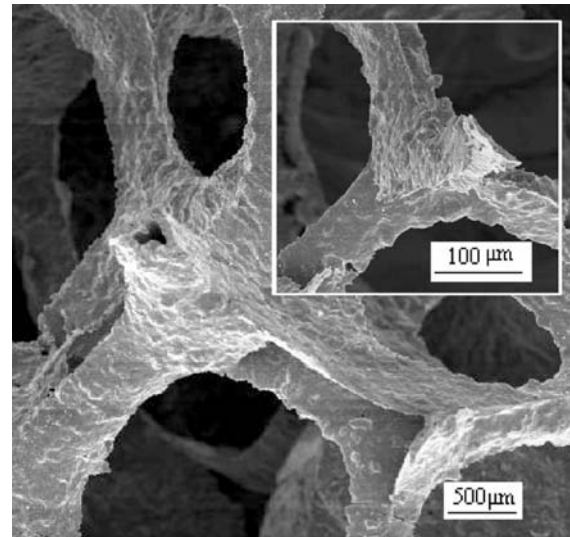


Fig. 4 Microstructure of Si₃N₄ porous ceramic skeleton (SEM)

Table 3 The parameters of porous ceramic skeleton

The number of pore (ppi)	Pore size (<i>d</i> , mm)	Struts width (<i>t</i> , mm)	Struts length (<i>l</i> , mm)
10 ± 2	3 ± 0.5	2.7 ± 0.5	7 ± 1

Table 4 The density and porosity of porous ceramic skeleton

Density of porous solid (g/cm ³)	Density of solid material (g/cm ³)	Relative density (g/cm ³)	Calculation result (g/cm ³)	Open porosity (%)
0.581 ± 0.05	3.17 ± 0.05	0.186 ± 0.02	0.185	81.4 ± 2

$$\frac{\rho^*}{\rho_s} = C \left(\frac{t}{l}\right)^2 \left[1 - D \left(\frac{t}{l}\right)\right] \quad (I)$$

where ρ^* is the density of porous solid, ρ_s is the density of the fully dense solid material, C , D is the constant, $C = 1.2$, $D = 0.7$ [17]. The density and porosity of porous ceramic skeleton are shown in Table 4.

The sintering additives as Al powder would be melted gradually with the increasing sintering temperature above 700 °C during the fabrication process of porous ceramic. The liquid Al at elevated temperature was diffused in the solid phases, thereby removing the cracks and jamming the hollow struts in the ceramic foam (as shown in Fig. 3). The additives of Al₂O₃ and ZrO₂ and in situ SiO₂ produced a glass phase at a higher temperature above 1300 °C. Owing to the existing of the additives, there should be few cracks in the struts, and the central void itself would lose the sharp

edged features typical of other reticulated foams as shown in Fig. 5a. Central void has a round appearance as shown in Fig. 5b and the hollow struts were jammed as shown in Fig. 5c.

Si₃N₄/Mg composites ceramic/metal interface

It is a well known fact that interfacial reactions take place in most of the ceramic–metal systems. For example, In the composite Al₂O₃–Fe, The iron aluminate spinel formation takes place at temperatures above 1000 °C [5]. In Al₂O₃/Al–Mg composites, Mg reacts with alumina, and MgAl₂O₄ spinel forms at the ceramic/metal interface [18]. A similar situation is observed between Mg and Si₃N₄. It can take place during the process of pressureless infiltration of melted magnesium into the porous Si₃N₄ ceramic network structure. In Si₃N₄/Mg

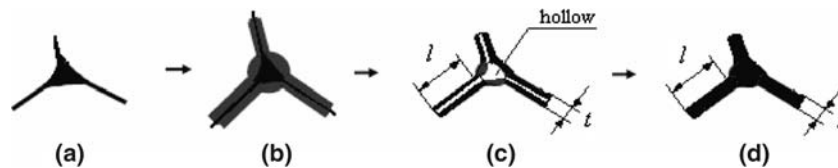
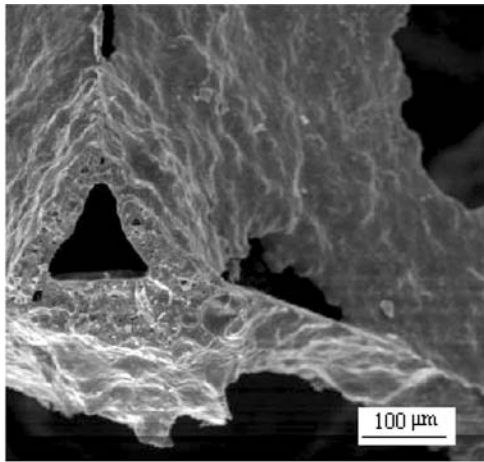
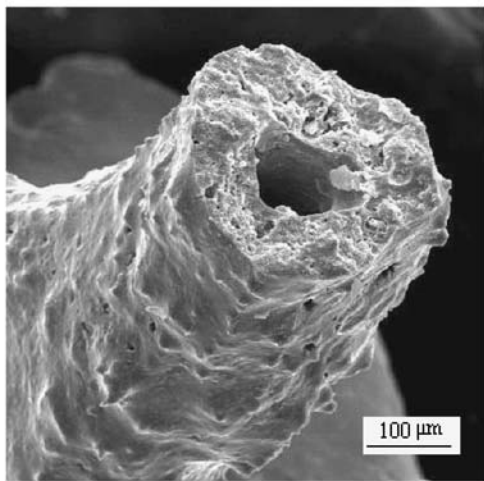


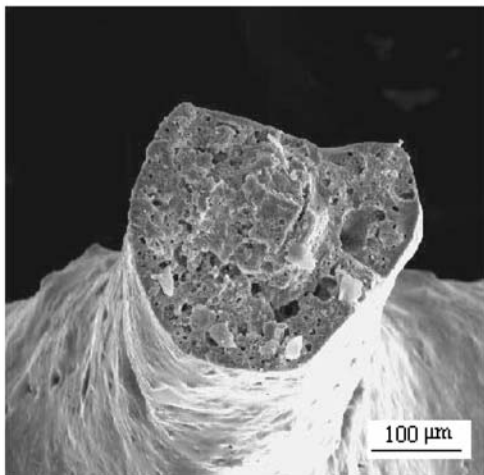
Fig. 3 The fabrication process and the struts parameters of porous ceramic. (a) Polymer strut, (b) polymer strut coated with ceramic slurry, (c) ceramic skeleton with hollow struts, (d) ceramic skeleton with jammed struts.



(a) The hollow struts with a sharp edged features



(b) The hollow struts with a round appearance

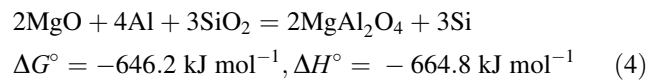
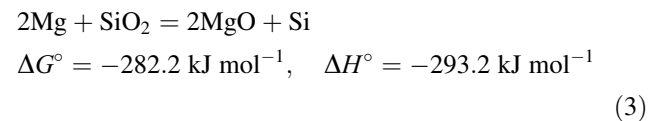
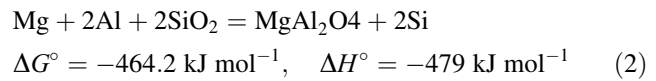
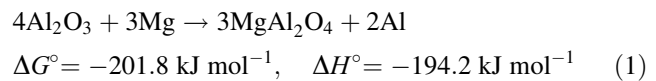


(c) The jammed struts of porous ceramic

Fig. 5 Microstructure of struts of Si_3N_4 porous ceramic skeleton. (a) The hollow struts with a sharp edged features, (b) the hollow struts with a round appearance, (c) the jammed struts of porous ceramic

composites, the strut surface of Si_3N_4 ceramic formed an oxide film of Al_2O_3 and SiO_2 (Fig. 6a) due to the ceramic materials sintered at oxidizing atmosphere, as was shown by the energy spectrum analyses (ESA, OXFORD INCA) in Fig. 6b. The result was confirmed by XRD analyses as shown in Fig. 7. In this experiment, the sintered ceramic powder was attained by crushing the porous ceramic skeleton.

It was indicated that the ceramic/metal interface contained rich Mg, Al, Si and O, which formed interfacial reaction product MgAl_2O_4 spinel. The interface reactions are shown as follow:



On the one hand, the above reactions (1)–(4) are expected to take place ($\Delta G < 0$) since MgAl_2O_4 is a thermodynamically stable phase in the Al–Mg–O system. The paper [19] confirmed that when Mg composition is above 0.1 wt%, the reactions would occur spontaneously. The Mg content used in this study is much more rich and the reaction terminated when the count of Al_2O_3 and SiO_2 deplete entirely. The growth of spinel layer is controlled by the counter diffusion of Mg^{2+} ions [18]. On the other hand, reactions (1)–(4) are exothermic reactions ($\Delta H < 0$), the heat quantity decreases turn slowly in infiltration front and this redound to pressureless infiltration. On the ceramic/metal interface a layer of the spinel phase is visible (Fig. 8). The thickness of the spinel layer was between 500 and 6000 nm, literature [14] has got the similar conclusion.

The spinel phase at the ceramic/metal interface plays a significant role in determining the mechanical properties of 3DNMMCs, such as compressive strength, tensile strength and wear resistance. This attributes to the thickness and morphology of the spinel. Above viewpoint has already been discussed in the literature. Some authors report that a

Fig. 6 SEM image of the cross-section microstructure of Si₃N₄ porous ceramic showing the oxide film (Al₂O₃ and SiO₂)

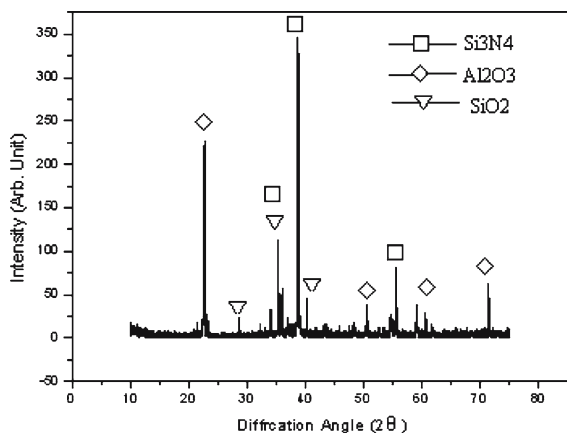
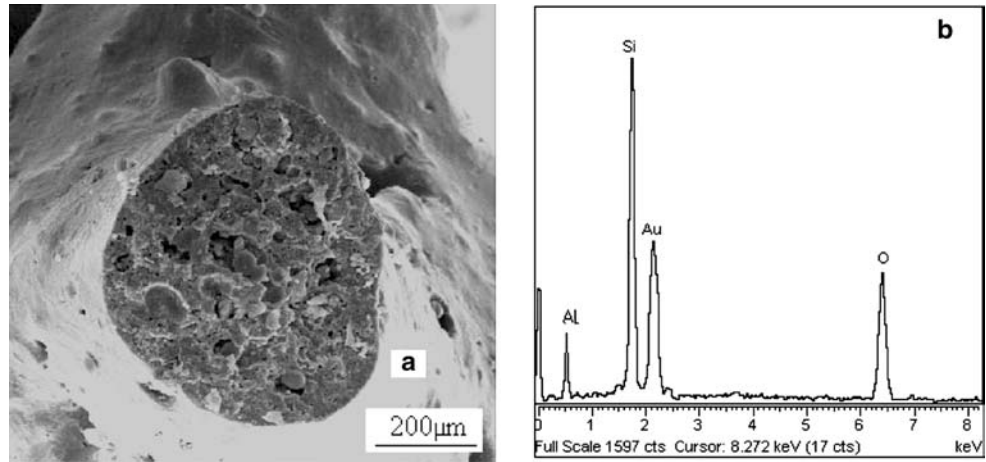


Fig. 7 XRD analyses of the strut surface of Si₃N₄ porous ceramic

thin spinel layer at the ceramic/metal interface plays a good and beneficial role in MMCs [14, 20]. Other authors give an inverted conclusion [21, 22]. The relations between compression stress and strain as magnesium alloy un-reinforced by ceramic form (specimen A) and two different composites reinforced by Si₃N₄ ceramic foam with different interface character (specimen B and C) are studied in this paper (which are shown in Fig. 9). Figure 10 shows the different interface microstructure of different composites. The interface of 3DNMMCs shown in Fig. 10a is worse and wider than that of 3DNMMCs shown in Fig. 10b. The width of interface in Fig. 10a is 6000 nm, whereas the width of interface in Fig. 10b is 500 nm. Comparatively, 3DNMMCs of Fig. 10b has more excellent interface structure characteristics. Interface structure characteristics such as interface width, interface crack and interface debonding are the main failure mechanism of composites. From Fig. 10a (bending arrow), the interface debonding and crack were observed distinctly whereas that in Fig. 10b is not only thin but also compact. So the composites providing with the interface in Fig. 10b were affirmed to be excellent than that of in Fig. 10a.

The values of the elastic modulus $E_A = 45$ GPa, $E_{RB} = 58$ GPa, $E_{RC} = 60$ GPa, E_A as the elastic modulus of magnesium alloy (specimen A), E_{RB} as the elastic modulus of specimen B and E_{RC} as the elastic modulus of specimen C. The values of rupture modulus $\sigma_{fA} = 240$ MPa, $\sigma_{fRB} = 250$ MPa, $\sigma_{fRC} = 255$ MPa, σ_{fA} as the rupture modulus of magnesium alloy, σ_{fRB} as the rupture modulus of specimen B and σ_{fRC} as the rupture modulus of specimen C. The un-reinforced magnesium alloys show a high elongation without necking of more than 25% at room temperature on the condition of enduring the compressive stress. The composites have a shorter elongation of 10% at room temperature. The composite (b) has a higher compression stress than composite (a) at the same strain although they have the same elongation. Other properties such as tensile and elongation with different interface are shown in Table 2.

Conclusions

One Si₃N₄/Mg composite reinforced by ceramic interpenetrating network structure has been fabricated via pressureless infiltration technology. The porous ceramic skeleton had a special microstructure with a hollow strut, which was padded by the sintering additives. Some interface reactions occurred on the ceramic/metal interface due to the strut surface of porous ceramic covered with a thin oxide film. The oxide film is proved as Al₂O₃ and SiO₂ by SEM, XRD and ESA. A thin MgAl₂O₄ spinel layer is observed in the present work. The special microstructure of ceramic–metal composites with an interpenetrating network, which creates their special properties, encourages to work on the synthesis of these materials and to investigate their microstructure and properties. Especially, the composites with a thin spinel layer located on the ceramic/metal interface possess excellent properties. It will give an opportunity to develop an interpenetrating network of ceramic–metal composites.

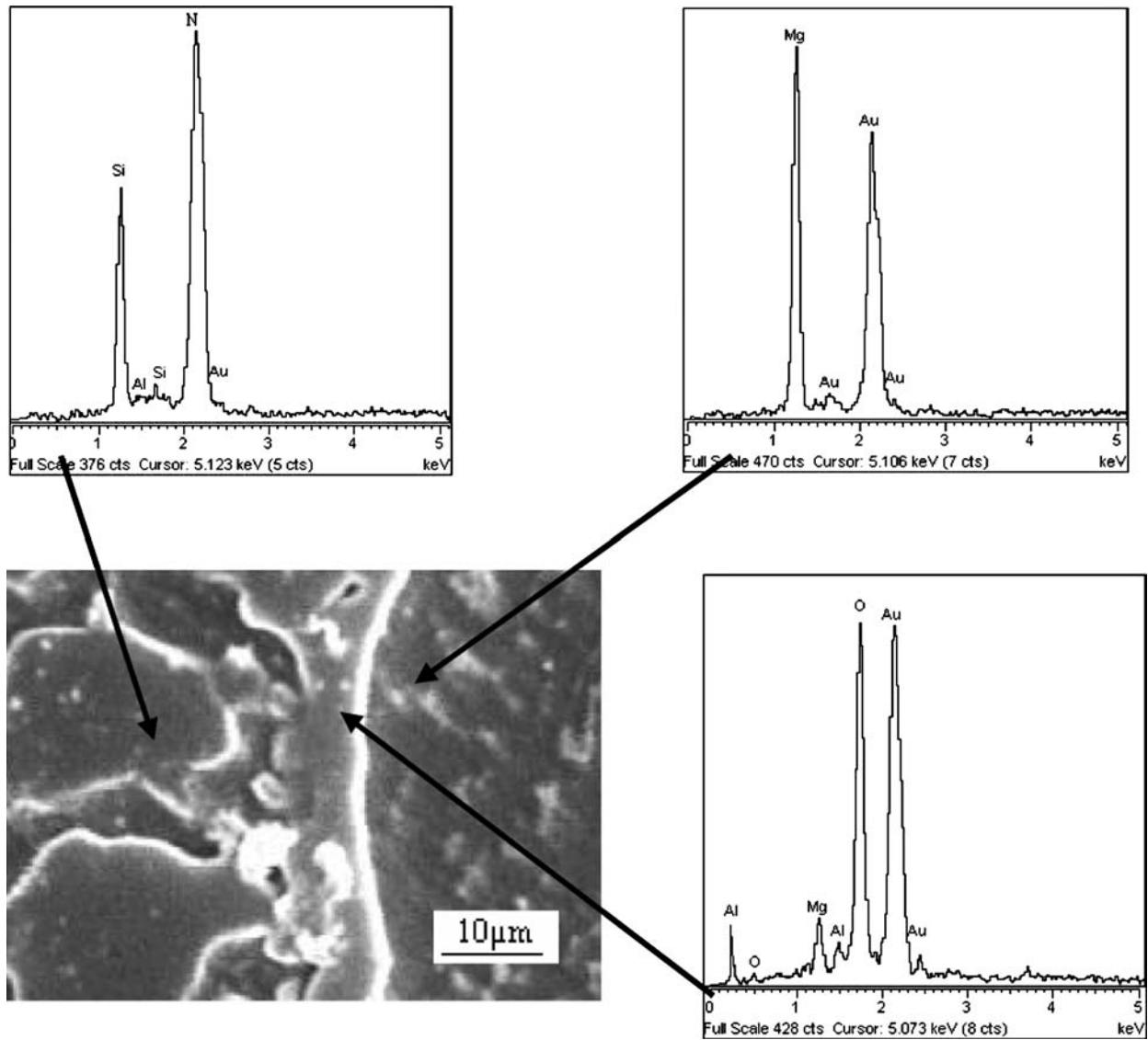


Fig. 8 SEM image of the microstructure of $\text{Si}_3\text{N}_4/\text{Mg}$ composites showing the distribution of the MgAl_2O_4 spinel

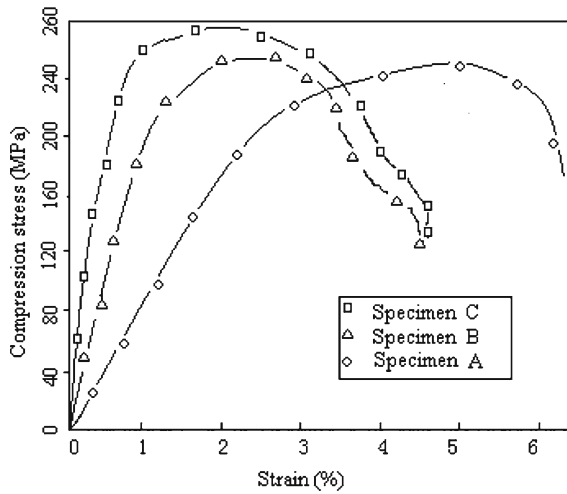
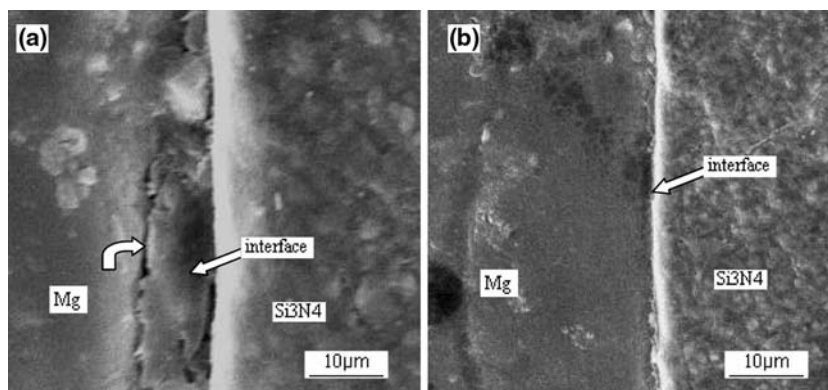


Fig. 9 The compression stress–strain relations of alloy and composites

Fig. 10 The different interface microstructure of $\text{Si}_3\text{N}_4/\text{Mg}$ composites



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